# Intramolecular Diels-Alder Reaction of a Sulphonyl-activated 1,3,9-Triene

# Andreas Weichert and H. Martin R. Hoffmann\*

Department of Organic Chemistry, University of Hannover, Schneiderberg 1 B, D-3000 Hannover, Federal Republic of Germany

The eudesmane precursor (6) containing a *trans* ring fusion has been prepared stereoselectively by an inverse electron-demand, intramolecular  $(4 + 2) \pi$  cycloaddition with phenylsulphonyl as controlling group.

The intramolecular Diels–Alder reaction (IMDA) has attracted continued interest as a method for the regio- and stereo-selective construction of molecules with several chiral centres.<sup>1</sup> Hitherto,  $\pi$ -components which are activated by sulphonyl substituents have hardly been studied, although recently Craig *et al.*<sup>2a</sup> and Battiste and his co-workers<sup>2b</sup> have described IMDA reactions of vinylic sulphones.

In an approach towards *trans* fused octahydronaphthalenes of the eudesmane type  $3^{a}$  we have investigated an IMDA reaction of a dienyl sulphone, *i.e.* a Diels-Alder cyclization with inverse electron-demand. Preparation of a suitably functionalized 1,3,9-triene began with isopulegol (1), which was cleaved oxidatively to the aldehyde (2).<sup>3b</sup> Coupling of (2) with phenyl vinyl sulphone in the presence of DABCO (1,4diazabicyclo[2.2.2]octane)<sup>4.5</sup> afforded the allylic alcohol (4a) which was mesylated and subjected to elimination-cyclization [(4a) $\rightarrow$ (5) $\rightarrow$ (6)]. Flash chromatography (ether-petroleum, 2:1) yielded an oily mixture of four diastereoisomers (4 acetoxycarbonyl signals in the  ${}^{13}$ C NMR spectrum) in reproducible fashion. 400 MHz <sup>1</sup>H NMR showed two pairs of isomers which were epimeric at C-7 and C-10, respectively (Table).

A priori, 8 diastereoisomers ( $2^4$  stereoisomers) are possible for a decalin with 4 chiral centres. Since only 4 diastereoisomers were formed, the ring junction had to be either *cis* or *trans*. Fractional crystallization (Et<sub>2</sub>O) yielded the major isomer, m.p. 189 °C, which was submitted to X-ray crystal structure analysis (Figure 1), showing *trans* ring fusion. Therefore, the 3 remaining diastereoisomers were also *trans* fused decalins.

Clearly, the preferred transition state is *endo* with respect to the olefinic methyl group, which becomes angular or axial in the product (Figure 2). Previously, Wilson,<sup>6</sup> and Taber<sup>7</sup> have studied related IMDA reactions of 1,3,9-trienes with an olefinic methyl group at C-9 [as is also present in our triene (5)] and a second methyl group at C-3. In this case, high *trans* stereoselectivity has been ascribed to steric bulk at C-3.<sup>6-8</sup> The triene (5) lacks the 3-methyl group and has a bulky phenyl-sulphonyl group at the more remote C-2. In our case, the angular methyl group appears to control the steric outcome, possibly in combination with an electronically altered

Table. Characteristic 400 MHz <sup>1</sup>H NMR data<sup>*a.b*</sup> of (6).

Diastereoisomer <sup>c</sup>	Ratio <sup>d</sup>	2-H <sup>d</sup>	7-H	14-H	11 <b>-H</b>	12-H
( <b>6</b> ) (7 <i>a</i> ,10 <i>a</i> )	1	6.88	4.54 (dd 11/5)e	2.01	0.97	0.83
(7α,10β)	2	(III) 7.04	(dd, 11/5) 4.54 (dd, 11/5)	2.02	(d, 7.3) <sup>2</sup> 1.04 (d, 6, 5)#	(s) 0.75 (s)
(7β,10α)	2	(m) 7.26	(dd, 11/5) <sup>2</sup> 4.69	2.07	1.07	(s) 0.70
(7β,10β)	4	(m) 7.14 (m)	(dd, 2.5) <sup>3</sup> 4.73 (dd, 2.5) <sup>f</sup>	(s) 2.04 (s)	(d, 7.0) <sup>s</sup> 1.08 (d, 6.0) <sup>g</sup>	(8) 0.76 (8)
		(m)	$(aa, 2.5)^{3}$	(s)	(a, 6.0)*	(S)

<sup>a</sup> Bruker instrument. <sup>d</sup> Determined in CDCl<sub>3</sub> with Me<sub>4</sub>Si as internal standard; coupling constants <sup>3</sup>J in Hz. <sup>c</sup> Diastereoisomeric at C-7 and C-10; cf. formula (6). <sup>d</sup> The intensity of the well resolved 2-H signal gave the diastereoisomeric ratio. <sup>e</sup> The large coupling constant (11 Hz) demands an axial proton at C-7, which is coupled to the *trans* diaxial proton at C-8 [cf. (6) ( $7\alpha$ ,  $10\alpha$ ); (6) ( $7\alpha$ ,  $10\beta$ )]. <sup>f</sup> The equatorial proton is coupled equally with the 8-methylene protons (dd, apparent triplet) [cf. (6) ( $7\beta$ ,  $10\alpha$ ); (6) ( $7\beta$ ,  $10\beta$ )]. <sup>g</sup> The smaller <sup>3</sup>J coupling constant is observed for the methyl group which is equatorial at C-10, the larger one for the axial epimer (cf. T. M. Moynehan, K. Schofield, R. A. Y. Jones, and A. R. Katritzky, J. Chem. Soc., 1962, 2637; J. A. Marshall and J. J. Partridge, *Tetrahedron*, 1969, **25**, 2159).



Figure 1. X-Ray crystal structure of (6) (major isomer 7β,10β).



Scheme. Reagents and conditions: i, Pb(OAc)<sub>4</sub>, benzene, reflux 2 h; ii, CH<sub>2</sub>=CHSO<sub>2</sub>Ph (3), DABCO, room temp., 8 weeks; iii, MsCl, EtNPr<sup>1</sup><sub>2</sub>, catalyst DMAP, -20 °C, 20 h; iv, pyridine, toluene, 170 °C, 24 h, sealed tube.

transition state due to the electron-attracting phenylsulphonyl group.<sup>9</sup>

## Experimental

For the preparation and spectral properties of 8-acetoxy-2-phenylsulphonyl-5,9-dimethyl-3-hydroxydeca-1,9-diene (4a) cf. ref. 4a.

8-Acetoxy-5,9-dimethyl-3-methylsulphonyloxy-2-phenylsulphonyldeca-1,9-diene (4b).—Methanesulphonyl chloride (0.55 g, 4.73 mmol) in dry  $CH_2Cl_2$  (6 ml) was added to a solution of the allylic alcohol (4a) (1.5 g, 3.94 mmol), ethyldiisopropylamine (1.53 g, 11.8 mmol), and a catalytic amount of *p*-dimethylaminopyridine (DMAP) in CH<sub>2</sub>Cl<sub>2</sub> (6 ml) at -20 °C and the mixture was stirred for 20 h under nitrogen. The solution was then poured into ice-cooled 1M HCl and the organic layer was separated, washed with saturated aqueous NaHCO<sub>3</sub> and water, dried (MgSO<sub>4</sub>), filtered, and evaporated under reduced pressure. The crude brown oil was purified by flash chromatography (ether-light petroleum, 5:1) on silica gel to give the mesylate (4b) as a colourless oil (1.34 g, 74%) (diastereoisomeric mixture) (Found:  $M^+$  – COMe, 415.1248. C<sub>21</sub>H<sub>30</sub>O<sub>7</sub>S<sub>2</sub> requires *M*, 415.1249); v<sub>max</sub>(CHCl<sub>3</sub>) 1 730 (C=O), 1 370 and 1 175 (OSO<sub>2</sub>CH<sub>3</sub>), and 1 320 and 1 145 cm<sup>-1</sup> (SO<sub>2</sub>Ph); δ<sub>H</sub>(200 MHz, CDCl<sub>3</sub>) 7.96–7.54 (5 H, m, Ar), 6.63/6.61 (1 H, d, <sup>2</sup>J 1.5 Hz, 1-H), 6.26 (1 H, br s, 1-H), 5.26 [1 H, m, C(H)OAc], 5.13/5.08 [1 H, t, <sup>3</sup>J7 Hz, C(H)OMs], 4.91 (2 H, m, =CH<sub>2</sub>), 2.76/2.74 (3 H, OSO<sub>2</sub>Me), 2.08/2.05 (3 H, s, COCH<sub>3</sub>), 1.73/1.69 (3 H, s, CCH<sub>3</sub>), 1.67–1.01 (7 H, m), and 0.87/0.82 (3 H, d, <sup>3</sup>J 6 Hz, CHCH<sub>3</sub>); δ<sub>C</sub>(50 MHz, CDCl<sub>3</sub>) 170.22/170.16 (s, C=O), 150.49/149.97 (s, C-2), 143.12/143.01 (s, C=CH<sub>2</sub>), 139.42 (s, Ar), 134.31, 129.63, 128.61 (d, Ar), 127.41/126.68 (t, C-1), 112.93/112.70 (t, C=CH<sub>2</sub>), 77.26/77.21, 75.58/75.42 (d, CHOMs, CHOAc), 44.16/43.96 (t, C-4), 38.56/38.47 (q, OSO<sub>2</sub>Me), 32.61/31.02 (t, C-7), 29.75/29.58 (d, CHCH<sub>3</sub>, 29.12/28.91 (t, C-6), 21.12/19.60 (q, C-12), 19.41/19.03 (q, CHCH<sub>3</sub>); m/z (220 °C) 458 (1%,  $M^+$ ), 415 (14), 320 (32), 302 (17), 161 (100), 125 (56), 109 (82), 94 (30), 78 (45), and 43 (54).

For spectral data of 8-acetoxy-5,9-dimethyl-2-phenyl-sulphonyldeca-1,3,9-triene (5) cf. ref. 4a.

7B-Acetoxy-6,10B-dimethyl-3-phenylsulphonylbicyclo[4.4.0]dec-2-ene (6) (Major Isomer).-Dry pyridine (25 ml) was added to the mesylate (4b) (2.0 g, 4.37 mmol) in dry toluene (290 ml, 0.015m) and the solution was heated at 170 °C for 24 h in a sealed tube. After dilution with ether, the reaction mixture was washed with 1M HCl solution and water. The organic layer was dried (MgSO<sub>4</sub>), filtered, and concentrated. The residue when subjected to flash chromatography (silica gel, ether-light petroleum, 2:1) afforded an oil (1.34 g, 85%) (diastereoisomeric mixture, 4:2:2:1, cf. Table), from which the major isomer 78,108 (0.59 g, 37%) crystallized selectively (m.p. 189 °C) upon addition of ether (Found: C, 66.3; H, 7.25. C<sub>20</sub>H<sub>26</sub>O<sub>4</sub>S requires C, 66.27; H, 7.23%) (Found:  $M^+$  362.1552.  $C_{20}H_{26}O_4S$  requires 362.1552); v<sub>max</sub>(CHCl<sub>3</sub>) 3 020, 2 960, 2 940, 1 725, 1 445, 1 370, 1 315, 1 305, 1 250, 1 150, 1 090, and 910 cm<sup>-1</sup>;  $\delta_{\rm H}$ (300 MHz, CDCl<sub>2</sub>) 8.05-7.50 (5 H, m, Ar), 7.14 (1 H, m, =CH), 4.73 (1 H, dd, <sup>3</sup>J 2.5 Hz, CHOAc), 2.36–1.26 (13 H, m), 2.04 (3 H, s, COCH<sub>3</sub>), 1.08 (3 H, d,  ${}^{3}J$  6 Hz, CHCH<sub>3</sub>), and 0.76 (3 H, s, CCH<sub>3</sub>);  $\delta_{C}$ (75 MHz, CDCl<sub>3</sub>) 170.38 (s, C=O), 139.51 (s, C=CH), 138.96 (d, C=CH), 138.74 (s, Ar), 133.18, 129.14, 128.02 (d, Ar), 75.90 (d, CHOAc), 45.17 (d, =CHCH), 36.12 (s, CCH<sub>3</sub>), 31.22, 30.56 (t, C-4, C-8), 29.19 (d, CHCH<sub>3</sub>), 25.87, 20.84 (t, C-5, C-9), 21.15 (q, COCH<sub>3</sub>), 19.47 (q, CCH<sub>3</sub>), and 16.32 (q, CHCH<sub>3</sub>); m/z $(100 \text{ °C}) 362 (8\%, M^+), 320 (5), 303 (90), 161 (100), 145 (22),$ 125 (34), 104 (30), 91 (22), 78 (28), and 43 (44).

### Acknowledgements

We thank Alexandra M. Z. Slawin and Dr. D. J. Williams of Imperial College London for the X-ray structure determination. Our work was supported by the Deutsche Forschungsgemeinschaft and Fonds der Chemischen Industrie.

#### References

- (a) D. Craig, Chem. Soc. Rev., 1987, 16, 187; (b) A. G. Fallis, Can. J. Chem., 1984, 62, 183; (c) E. Ciganek, Org. React., 1984, 32, 1; (d) D. F. Taber, 'Intramolecular Diels-Alder and Ene Reactions,' Springer Verlag, New York, 1984; (e) G. Brieger and J. N. Bennett, Chem. Rev., 1980, 80, 63; (f) R. L. Funk and K. P. C. Vollhardt, Chem. Soc. Rev., 1980, 9, 41; (g) W. Oppolzer, Synthesis, 1978, 798; (h) W. Oppolzer, Angew. Chem., Int. Ed. Engl., 1977, 16, 10.
- 2 (a) D. Craig, D. A. Fischer, Ö. Kemal, and T. Plessner, *Tetrahedron Lett.*, 1988, **29**, 6369; (b) L. Strekowski, S. Kong, and M. A. Battiste, J. Org. Chem., 1988, **53**, 901.
- 3 (a) Review: C. H. Heathcock in 'The Total Synthesis of Natural Products,' vol. 5, ed. J. ApSimon, Wiley, New York 1983; (b) K. Shankaran and A. S. Rao, *Indian J. Chem., Sect. B*, 1979, **18**, 507.
- 4 (a) H. M. R. Hoffmann, A. Weichert, A. M. Z. Slawin, and D. J. Williams, *Tetrahedron*, 1990, 46, in press; (b) P. Auvray, P. Knochel, and J. F. Normant, *Tetrahedron Lett.*, 1986, 27, 5095; P. Auvray, P. Knochel, and J. F. Normant, *Tetrahedron*, 1988, 44, 6095.
- 5 (a) H. M. R. Hoffmann and J. Rabe, Angew. Chem., Int. Ed. Engl., 1983, 22, 795; (b) J. Rabe and H. M. R. Hoffmann, *ibid.*, 1983, 22, 796 and subsequent publications. Review: S. E. Drewes and G. H. P. Roos, *Tetrahedron*, 1988, 44, 4653.
- 6 S. R. Wilson and D. T. Mao, J. Am. Chem. Soc., 1978, 100, 6289.
- 7 D. F. Taber and S. A. Saleh, Tetrahedron Lett., 1982, 23, 2361.
- 8 Ref. 1a, p. 221; ref. 1d, p. 48.
- 9 For another recent example of stereocontrol due to a phenylsulphonyl group see ref. 4a.

Paper 0/00965B Received 2nd March 1990 Accepted 5th April 1990